

# Statistical *Ab Initio* Analysis of Electron Trapping Oxide Defects in the Si/SiO<sub>2</sub> Network

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**Abstract**—It is well established that oxide defects are a major threat concerning the reliability of electronic devices. Point defects like the oxygen vacancy and hydrogen related defects like the hydrogen bridge and the hydroxyl- $E'$  center can trap charges from the Si substrate during operation. Here we present a statistical study of parameters by analyzing 144 defects of each type in amorphous SiO<sub>2</sub>. We use density functional theory in conjunction with a hybrid functional to provide a large and accurate dataset of defect parameters such as the relaxation energy or the charge transition level and link them to bias temperature instability in nMOSFET devices.

**Index Terms**—BTI, nMOS, SiO<sub>2</sub>, hydroxyl- $E'$

## I. INTRODUCTION

Point defects have received significant interest over the past decades due to their various implications related to reliability issues [1]. For example, oxide defects in metal oxide semiconductor field effect transistor (MOSFET) devices are suspected to be responsible for bias temperature instability (BTI) [2] and random telegraph noise (RTN) [3] by capturing charges from the substrate during operation. The involved charge transition mechanisms causing BTI can be described by a nonradiative multi-phonon (NMP) process [4]. However, experimental and theoretical studies are still challenging due to the amorphous nature of the oxide, as properties of defects vary over a broad range and can thus only be treated in a statistical manner. Although HfO<sub>2</sub> is more and more used as the gate oxide in modern MOSFETs, even in ultra-scaled devices, an interfacial layer of SiO<sub>2</sub> persists to ensure good lattice matching to Si [5] and thus defect studies in the Si/SiO<sub>2</sub> network remain highly relevant as these defects are closest to the channel and therefore dominate the observed charge trapping.

Among the various defect types, the oxygen vacancy (OV) is the most investigated defect candidate in a-SiO<sub>2</sub>. Studies have clearly shown, however, that the OVs charge transition levels are far outside a reasonable interaction window regarding the charge exchange with carriers in the Si substrate [6]. Nevertheless, hydrogen atoms, which are deliberately introduced into the device during annealing [7], can easily diffuse through the oxide and interact with already existing OVs to form a new defect type, the so-called hydrogen bridge (HB), with its trap levels near the band edges of Si. Another previously discovered defect candidate is the hydroxyl- $E'$  center (H- $E'$ ) [8], where a hydrogen atom breaks a strained Si-O bond

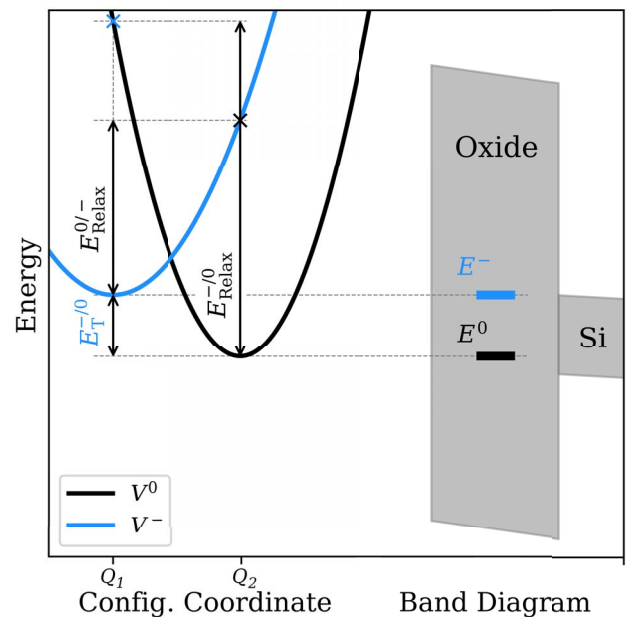


Fig. 1. Schematics of the potential energy surfaces (PES)  $V^0$  and  $V^-$  in the harmonic approximation for an oxide defect in two charge states 0 and  $-$  as a function of the configuration coordinate (left). The energy levels are shown within the context of the band diagram of a Si/SiO<sub>2</sub> network (right), with the minimum of the negative PES,  $E^-$ , assumed as the energy reference fixed at the conduction band edge. The energy difference  $E^- - E^0$  is defined as the thermodynamic trap level  $E_T^{-/0}$ .

in SiO<sub>2</sub>, forming a Si dangling bond facing a hydroxyl group. Even though these defects and their effect on pMOS device reliability have been investigated over the last years [9], a consistent theoretical study of their electron trapping properties is still missing. Here we use density functional theory (DFT) employing a hybrid functional to calculate properties for neutral and negatively charged defects on a statistical level. The initial, amorphous SiO<sub>2</sub> structures were created by performing a melt-and-quench procedure using a molecular dynamics algorithm and consecutive DFT cell optimizations. We statistically analyze charge transitions at the defect site by calculating the relaxation energies and charge transition levels of 144 initial defects of each kind introduced above as shown in Fig. 1. We show that the  $-/0$  charge transition

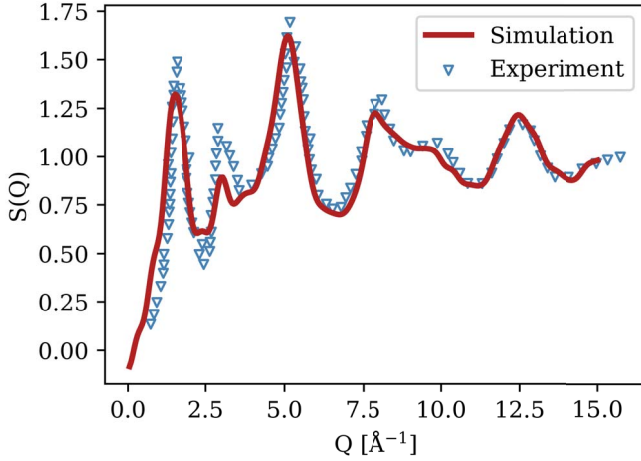


Fig. 2. Structure factor  $S$ , as a function of the scattering vector  $Q$  for the utilized structure used for defect calculations compared to experimental data [10].

levels of the hydrogen related defects are normally distributed in the vicinity of the conduction band edge of the Si substrate which renders them compatible to experimental investigations of RTN and PBTI in nMOS devices.

## II. METHODOLOGY

### A. DFT Setup

First-principle calculations were performed with periodic boundary conditions using the Gaussian plane wave (GPW) method as implemented in the CP2K code [11]. A double- $\zeta$  Gaussian basis set [12] was employed in combination with the Goedecker-Teter-Hutter (GTH) pseudopotentials [13] and an energy cutoff of 800 Ry. We use the non-local hybrid functional PBE0\_TC\_LRC [14] for all calculation to minimize the errors of electronic state calculations and set the convergence criteria for the SCF cycle to  $2.7 \times 10^{-6}$  eV. In addition, the auxiliary basis set pFIT3 [15] was employed to make calculations of the Hartree-Fock exchange, necessary for hybrid functionals, computationally more feasible. The Broyden-Fletcher-Goldfarb-Shanno (BFGS) algorithm was used to perform geometry optimizations in different charge states.

### B. Structure creation

We employ the melt-and-quench procedure [6] to the crystalline  $\text{SiO}_2$  polymorph  $\beta$ -cristobalite to create 20 structures of amorphous  $\text{SiO}_2$  with 216 atoms each.  $\beta$ -cristobalite was chosen since its density ( $2.33 \text{ g cm}^{-3}$ ) is closest to the experimental value of a- $\text{SiO}_2$  thin films ( $2.18 \text{ g cm}^{-3}$ ) [16] compared to other polymorphs like  $\alpha$ -Quartz ( $2.65 \text{ g cm}^{-3}$ ). The structures were created with a molecular dynamics (MD) algorithm as implemented in the LAMMPS code [17] with the classical ReaxFF potential [18]. This force field was successfully used in previous publications [19] [20] due to its ability to describe dynamic bond formation and breakage. For all MD calculations, the time step was set to 0.1 fs and the thermostat and barostat parameters were chosen to keep the root mean squared

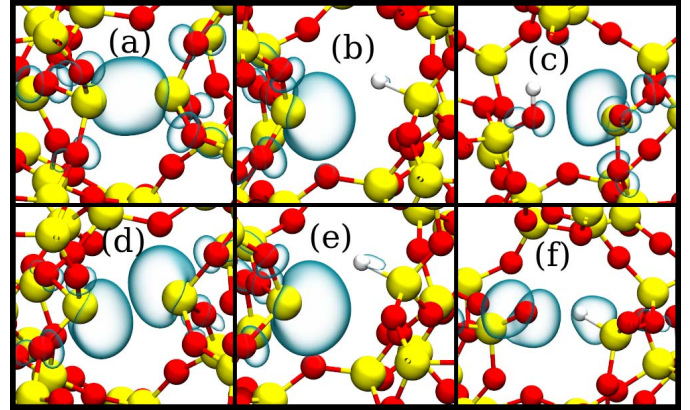


Fig. 3. Examples of the a- $\text{SiO}_2$  defect candidates, oxygen vacancy (left), hydrogen bridge (middle) and hydroxyl- $E'$  center (right), in the neutral (top) and negative (bottom) charge states with their HOMO localized at the defect site.

fluctuations below  $10^{-5}$ . The  $\beta$ -cristobalite was heated above its artificial melting point to 5000 K and equilibrated at this temperature for 1 ns and cooled down subsequently to 0 K at a cooling rate of  $6 \text{ K ps}^{-1}$ . Consequently, we performed cell optimizations on the quenched structures within DFT to lower the internal forces to  $0.025 \text{ eV/\AA}$  and to reduce the pressure to 0 GPa within a tolerance of 0.01 GPa.

### C. Structural Analysis

The structures created using the procedure described above resemble important characteristics of realistic a- $\text{SiO}_2$ . The bond lengths ( $1.63 \pm 0.015 \text{ \AA}$ ) and angles ( $\text{O-Si-O} = 109.47 \pm 4.261^\circ$  and  $\text{Si-O-Si} = 143.41 \pm 13.203^\circ$ ) agree well with high-energy X-ray and neutron diffraction data [21] and are in the same range as previous DFT calculations [20]. The mass density is also in very good agreement with experiments ( $\rho = 2.17 \text{ g cm}^{-3}$ ) [16].

As Bragg's law cannot be applied in amorphous materials, the structure factor [22] is preferably used in amorphous materials to characterize the short-, medium- and long-range order. The structure factor of our model structure used for defect calculations compares very well to neutron diffraction data [10] over the whole  $Q$  range as shown in Fig. 2.

## III. DEFECT CANDIDATES

The three defect candidates studied in this work, oxygen vacancy, hydrogen bridge and hydroxyl- $E'$  center, are shown in Fig. 3 (a-f) together with their corresponding charge distribution for the highest occupied molecular orbital (HOMO) for the neutral and negative charge state. A detailed description of these defects can be found in [23]. For all three defect types, puckered configurations are also discussed, where a silicon atom moves through the plane of its three adjacent O atoms and often back-bonds to a fourth O atom.

OVs were created by removing a single O atom from the structure. For the HBs, an O was replaced by an H atom, while for the H- $E'$  a H atom was placed in the vicinity of an O. 144 initial defect structures of each kind were created in this way

and geometry optimizations were performed on them using the DFT setup described in the previous section.

#### IV. RELAXATION ENERGIES

In the following, the relaxation energy is denoted as  $E_{\text{Relax}}^{0/-}$  and  $E_{\text{Relax}}^{-/0}$  for the transitions  $0 \rightarrow -$  and  $- \rightarrow 0$  respectively as shown in Fig. 1. To calculate  $E_{\text{Relax}}$  of the accumulated defects, single point calculations were performed at charge state  $-$  (0) on the relaxed configuration in the opposite charge state 0 ( $-$ ).  $E_{\text{Relax}}^{0/-}$ , for example, can afterwards simply be calculated as the energy difference to the relaxed configuration in the new charge state:

$$E_{\text{Relax}}^{0/-} = V^-(Q_2) - V^-(Q_1) \quad (1)$$

with  $Q_1$  and  $Q_2$  being the equilibrium positions of the defect configuration in charge state  $-$  and 0, respectively.

The obtained relaxation energies for each defect type are summarized in Fig. 4. The sample size of the  $E_{\text{Relax}}$  distributions differs because a certain number of the single point calculations did not meet the required convergence criteria. The majority of the relaxation energies of the oxygen vacancies is distributed below 2 eV. Only puckered OV's have relaxation energies  $> 3$  eV. For the HB,  $E_{\text{Relax}}$  for both charge transitions are asymmetrically distributed between 1 and 3 eV with a peak at around 1.5 eV.  $E_{\text{Relax}}$  of the H- $E'$  resembles a normal distribution for the  $0/-$  transition with the peak at 2 eV, while for the  $-/0$  transition the relaxation energies are between the energy range 1-2 eV for all analyzed defects.

#### V. CHARGE TRANSITION LEVELS

The charge transition level (CTL) can be calculated by comparing the formation energy of a defect in two charge states. The formation energy of a defect in charge state  $Q$  is given by

$$E_{\text{Form}}^Q(R) = E_{\text{tot}}^Q(R) - E_{\text{tot}}^{\text{bulk}} - \sum_i \mu_i n_i + Q E_F + E^{\text{cor}} \quad (2)$$

where  $E_{\text{tot}}^Q$  is the total energy of the system with the defect in charge state  $Q$ ,  $E_{\text{tot}}^{\text{bulk}}$  is the total energy of the defect-free neutral bulk,  $\mu_i n_i$  stands for the chemical energy needed to add or remove atoms of kind  $i$  to the bulk to create the defect,  $E_F$  is the Fermi level and  $E^{\text{cor}}$  is a correction term necessary for DFT calculations of charged systems within periodic boundary conditions. The Fermi level is calculated with respect to the valence band maximum  $E_F = E_{\text{VBM}} + \epsilon_F$ . Here,  $E_{\text{VBM}}$  is approximated as the highest occupied Kohn-Sham orbital of the defect-free bulk structure [24]. The correction term  $E^{\text{cor}}$  has been calculated to 0.35 eV for our structures within the image charge correction scheme of Makov and Payne [25].

The CTL corresponds to the Fermi level where the charged and the neutral formation energies are equal and thus both charge states are equally stable. It is a key characteristic of a defect, as the trap level with respect to the Fermi level determines the thermodynamically preferred charge state. The position of the trap level can be shifted by applying a gate bias due to the occurring electric field in the oxide and band

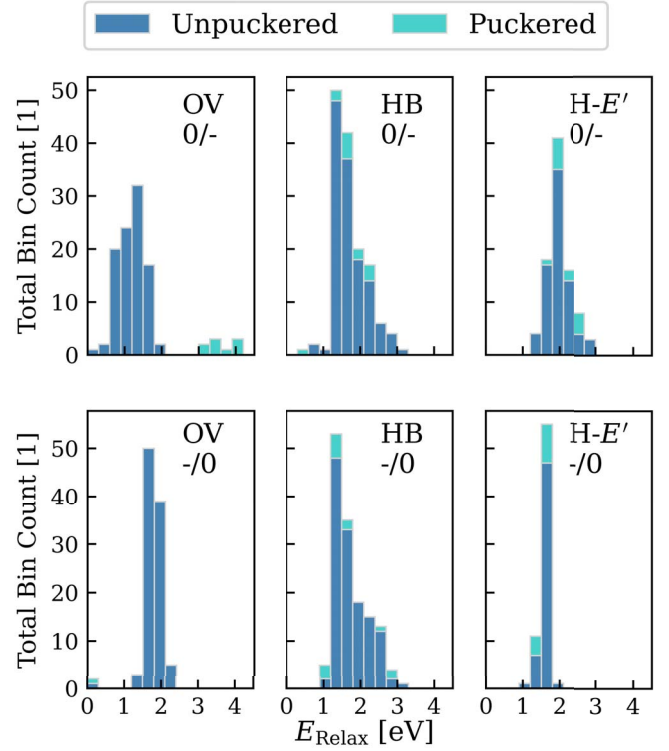


Fig. 4. Calculated relaxation energies for three defect types: Oxygen vacancy, hydrogen bridge and hydroxyl- $E'$  center.

bending near the interface. As the probability of the defect to capture charges from the substrate can be altered, the CTL is detectable experimentally and thus can be used to identify defects based on measurements. The CTL distributions in the band gap of  $\text{SiO}_2$  for electron capture for defect types OV, HB and H- $E'$  are shown in Fig. 5 with the calculated band gap of the Si bulk indicated as a gray area. The CTLs for all defect types resemble normal distributions with parameters given in Fig. 5. The majority of CTLs for the OV are above the Si conduction band edge. Only the puckered configurations possess lower CTLs. Nevertheless, their calculated formation energies are far higher compared to the unpuckered configurations and thus are less likely to occur in a- $\text{SiO}_2$ . In comparison to the CTLs of OV's for hole transitions [9] however, the CTLs for trapping electrons are closer to the energy of the relevant charge reservoir and the OV has to be considered as a defect candidate for trapping electrons from the substrate. For the HB and H- $E'$ , the CTLs are distributed around the conduction band edge of the Si substrate and thus these defects are plausible candidates for electron capture processes during operation [1].

#### VI. CONCLUSION

We provided a large and consistent data set of relaxation energies and charge transition levels for the  $-/0$  charge transition for the three defect candidates in amorphous  $\text{SiO}_2$ , oxygen vacancy (OV), hydrogen bridge (HB) and hydroxyl-

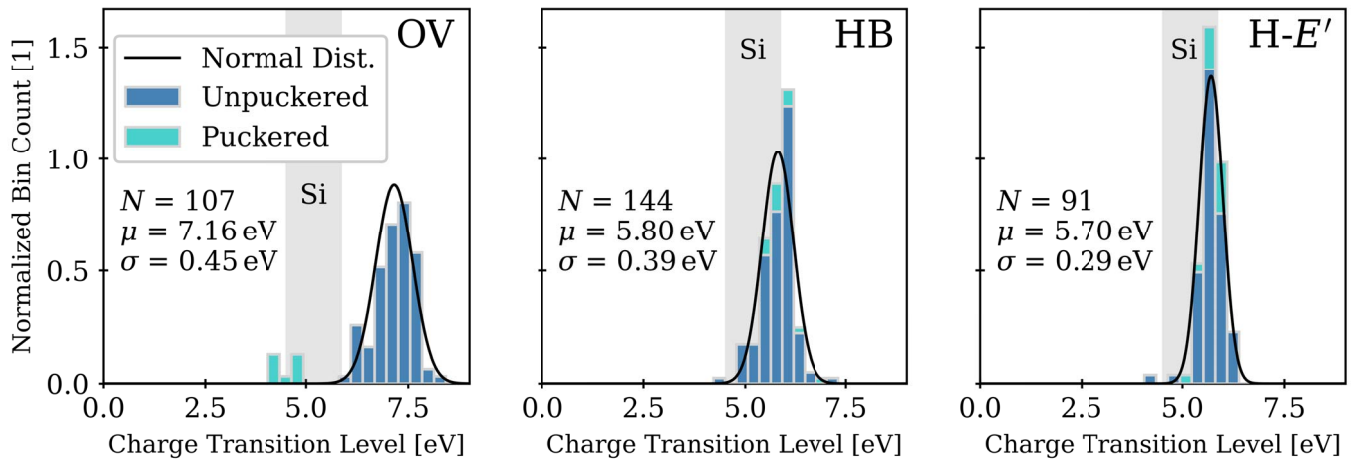


Fig. 5. Charge transition levels of defect types oxygen vacancy (left), hydrogen bridge (HB) and hydroxyl- $E'$  center ( $H-E'$ ) in a-SiO<sub>2</sub> for transition 0/-. A normal distribution was fitted to the data with the fitting parameters given in the plot. The Band gap of Si in a Si/SiO<sub>2</sub> network is indicated by a gray area.

$E'$  center ( $H-E'$ ), that can be directly linked to experimental characterizations in MOSFET devices. Our calculations show that the charge transition levels (CTLs) of OV's are distributed above the conduction band edge of the Si substrate. Contrary to the +/0 CTLs, some OV's possess trap levels within a reasonable range to interact with the carrier reservoir at high electric field conditions, potentially provoking the effect of bias temperature instability. On the other hand, CTLs of HBs and  $H-E'$ 's are distributed much closer to the Si conduction band edge of Si and hence are possible candidates to explain random telegraph noise under positive bias conditions.

#### ACKNOWLEDGMENT

This project was funded by the European Union's Horizon 2020 research and innovation program under grant agreement No. 871813, within the framework of the project Modeling Unconventional Nanoscaled Device FABrication (MUNDFAB) and the CDL for Single-Defect Spectroscopy in Semiconductor Devices (SDS).

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